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# HL Paper 3

Ethylamine,  $\text{C}_2\text{H}_5\text{NH}_2$ , is a weak base.

- a. Explain why ethylamine has basic properties. [1]
- b. The  $\text{p}K_{\text{b}}$  values at 298 K for diethylamine and triethylamine are given in Table 15 of the Data Booklet. Identify which of the two compounds is the stronger base and suggest a reason why it is stronger. [2]

Stronger base:

Reason:

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